



catalyseur h

l'acétate de benzyle c<sub>9</sub>h<sub>10</sub>o<sub>2</sub> ou ch<sub>3</sub>cooch<sub>2</sub>c<sub>6</sub>h<sub>5</sub> est l'ester de l'acide acétique et de l'alcool benzylique il constitue l'odeur de tête du jasmin et est présent dans l'huile essentielle de

la synthèse de l'acétate de benzyle odeur de jasmin est réalisée au laboratoire à partir d'acide éthanoïque acide acétique et d'alcool benzylique c'est une réaction lente et équilibrée de

synthèse de l'acétate de benzyle suivie par ccm déterminer les quantités introduites et faire un tableau d'avancement on se propose de synthétiser de l'acétate de benzyle à partir d'alcool benzylique et

mode opératoire dans une cuve à chromatographie on introduit sur une hauteur de 0,5 cm un éluant constitué d'un mélange de cyclohexane et d'éthanoate d'éthyle dans les proportions 1/4

nous allons synthétiser un ester l'éthanoate de benzyle ou acétate de benzyle par action de l'acide éthanoïque sur l'alcool benzylique cet ester constitue la note de tête de l'odeur du jasmin et est

recupérer la phase organique dans un erlenmeyer de 100ml pour sécher la phase organique verser 2 ou 3 spatules de sulfate de magnésium mgso<sub>4</sub> anhydre dans l'erlenmeyer

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